



Characterization of the Physicochemical, Structural, Thermal, and Properties of Zinc Chloride after the Consciousness Energy Healing Treatment

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Abstract: Zinc chloride is used as a source of zinc in various pharmaceutical/nutraceutical formulations. The aim of the current study was to investigate the impact of The Trivedi Effect[®] - Consciousness Energy Healing Treatment (Biofield Energy Treatment) on physical, structural, thermal, and behavioral properties of zinc chloride using PXRD, PSD, FT-IR, UV-vis, and DSC analysis. Zinc chloride was divided into two parts – one part was control, while another part was treated with The Trivedi Effect[®] - Consciousness Energy Healing Treatment remotely by twenty renowned Biofield Energy Healers and defined as the Biofield Energy Treated sample. A significant alteration of the crystallite size and relative intensities of the PXRD peaks was observed in The Trivedi Effect[®] treated sample compared with the control sample. The average crystallite size of the treated sample was significantly increased by 16.43% compared with the control sample. The particles size values of the treated sample at d_{10} was decreased by 2.42%, whereas at d_{50} , and d_{90} it was increased slightly by 0.43%, and 0.57%, respectively compared to the control sample. Therefore, the surface area of the treated sample was increased by 1.63% compared with the control sample. The FT-IR spectroscopic analysis revealed that Zn-Cl stretching in the control sample was found at 511 cm^{-1} , whereas it was shifted downward to 508 cm^{-1} in the treated sample. The UV-vis analysis exhibited that wavelength of the maximum absorbance (λ_{max}) of the control and treated samples were at 196.4 nm and 196.2 nm, respectively. The DSC analysis exhibited that the melting temperature was not altered significantly, while the decomposition temperature was increased by 0.45% in the treated sample compared to the control sample. The latent heat of fusion of the treated sample (402.09 J/g) was significantly increased by 29.78% compared with the control sample (309.83 J/g). Similarly, the enthalpy of decomposition of the treated sample (243.13 J/g) was significantly decreased by 61.18% compared with the control sample (626.37 J/g). This indicated that the thermal stability of the treated sample might be improved compared to the control sample. The current study anticipated that the Consciousness Energy Healing Treatment might lead to produce a new polymorphic form of zinc chloride, which would be more soluble, bioavailable, and thermally more stable compared with the control sample. Hence, the treated zinc chloride would be very useful to design better nutraceutical/pharmaceutical formulations that might offer better therapeutic response against immunological disorders, inflammatory diseases, stress, aging, cancer, etc.

Keywords: Zinc Chloride, Biofield Energy Healing Treatment, Consciousness Energy Healing Treatment, Biofield Energy Healers, The Trivedi Effect[®], PXRD, Particle size, DSC

1. Introduction

Zinc is an essential element in the human body, which regulates a broad spectrum of physiological functions. It plays an important role in the human body, which are involved in the nucleic acid, protein metabolism and for the production of energy. Zinc also maintains the structural integrity of the biological membranes resulting in their protection against oxidative injury [1-2]. It possesses various pharmacological activities include retino-protective, fertility enhancing, and putative antiviral activities [3]. Zinc is also used in the treatment of Wilson's disease [4]. It has also immunomodulatory and antioxidant activities [1]. Zinc also prevent the development of several deficiency symptoms include hypogeusia, parakeratosis, anorexia, geophagia, dysosmia, growth retardation, hypogonadism, etc. [5-7]. Recently, it has been reported that zinc had increased the phagocytic index, and leukocyte count which potentiate the immunomodulatory effect along with *Glycyrrhiza glabra* [8]. Zinc chloride ($ZnCl_2$) is a hygroscopic inorganic salt and highly soluble in water [9]. It is used in pharmaceutical industry as drug and diagnostic agent due to its powerful astringent and mild antiseptic properties [10]. It is also used in deodorant and mouth-wash preparations. Zinc chloride is used as dentin desensitizer *i.e.* reduces the sensitivity of the teeth to heat and cold, along with protein precipitation and insulin preparation [10, 11]. The literature reported that zinc chloride has also been used in the treatment of cancer as a destructive agent [12] and in the pluripotency maintenance of mouse embryonic stem cell through the regulation of STAT3 signaling pathway [13]. Zinc chloride can be used for the palliative treatment of canine and malignant skin wounds [14]. It has the reverse effect on the locomotor behavior of rats [15]. Therefore, zinc chloride was considered as one of the components in the novel proprietary herbomineral formulation for the source of zinc and can be used for the prevention and treatment of various human disorders.

An élan vital or vital force which contributes the 'life' is preserved by every living organisms and known as prana by the Hindus, *qi* or *chi* by the Chinese, and *ki* by the Japanese. This is believed to co-relate with the soul, spirit, and mind. This hypothetical vital force is considered as the Biofield Energy. The Biofield Energy is infinite, paradimensional and dynamic electromagnetic field surrounding the human body. It can freely flow between the human and environment that leads to the continuous movement or matter of energy [16, 17]. Thus, a human has the ability to harness energy from the earth, the "universal energy field" and transmit it to any living or nonliving object(s) around the globe. The objects always receive the energy and respond in a useful way. This process is known as Biofield Energy Healing Treatment [18, 19]. Biofield (Putative Energy Fields) based Energy

Therapies are used worldwide to promote health and healing. The National Center of Complementary and Integrative Health (NCCIH) has been recognized and accepted Biofield Energy Healing as a Complementary and Alternative Medicine (CAM) health care approach in addition to other therapies, medicines and practices such as natural products, deep breathing, yoga, Qi Gong, Tai Chi, chiropractic/osteopathic manipulation, meditation, special diets, massage, homeopathy, progressive relaxation, guided imagery, acupuncture, acupressure, relaxation techniques, healing touch, hypnotherapy, movement therapy, rolfing structural integration, pilates, mindfulness, Ayurvedic medicine, traditional Chinese herbs and medicines, naturopathy, aromatherapy, essential oils, cranial sacral therapy Reiki, and applied prayer (as is common in all religions, like Hinduism, Christianity, Buddhism and Judaism) [20]. Biofield Energy Treatment (The Trivedi Effect[®]) has been reported in numerous peer-reviewed science journals with significant outcomes in many scientific fields such as cancer research [21], altered antimicrobial sensitivity of pathogenic microbes in microbiology [22-24], biotechnology [25, 26], genetics [27, 28], changing the atomic structure in relation to various metals, ceramics, polymers and chemicals [29, 30], altered physical and chemical properties of pharmaceuticals [31, 32], nutraceuticals [33, 34], organic compounds [35-37], and improved overall growth and yield of plants in agricultural science [38, 39]. The physicochemical properties such as particle size, crystallite size, crystalline structure, surface area, and thermal properties of a drug have a vital role in bioavailability as well as the stability of the drug during processing, formulation, packaging, and storage [40, 41]. The particle size, surface area, crystalline nature, chemical and thermal behavior of an atom/ion might be altered by the Biofield Energy Healing Treatment (The Trivedi Effect[®]) through the possible mediation of neutrinos [42]. By considering all these aspects, powder X-ray diffraction (PXRD), particle size distribution analysis (PSD), Fourier transform infrared (FT-IR) spectrometry, ultraviolet-visible (UV-vis) spectroscopy, and differential scanning calorimetry (DSC) analytical techniques were conducted in this study for the characterization of physical, structural, and thermal properties of the Biofield Energy Treated and untreated zinc chloride.

2. Materials and Methods

2.1. Chemicals and Reagents

Zinc chloride was procured from Tokyo Chemical Industry Co., Ltd. (TCI), Japan. All other chemicals used in the experiment were of analytical grade available in India.

2.2. The Trivedi Effect[®] Energy of Consciousness Healing Treatment Strategies

Zinc chloride was one of the components of the new proprietary herbomineral formulation, developed by our research team and it was used *per se* as the test compound for the current study. The test compound was divided into two parts, one part of the test compound did not receive any sort of treatment and was defined as the untreated or control zinc chloride sample. The second part of the test compound was treated with The Trivedi Effect[®] - Energy of Consciousness Healing Treatment (Biofield Energy Treatment) by a group of twenty renowned Biofield Energy Healers remotely and was denoted as Biofield Energy Treated or The Trivedi Effect[®] sample. Fifteen Biofield Energy Healers were remotely located in the U.S.A., two in Canada, one in the UK, one in Australia, and one in Germany, while the test compound was located in the research laboratory of GVK Biosciences Pvt. Ltd., Hyderabad, India. This Biofield Energy Treatment was provided for 5 minutes through Healer's Unique Energy Transmission process remotely to the test compound under the laboratory conditions. None of the Biofield Energy Healers in this study visited the laboratory in person, nor had any contact with the compounds. Similarly, the control compound was subjected to "sham" healers for 5 minutes, under the same laboratory conditions. The sham healer did not have any knowledge about the Biofield Energy Treatment. After that, the Biofield Energy Treated and untreated samples were kept in similar sealed conditions and characterized thoroughly by PXRD, PSD, FT-IR, UV-visible spectroscopy, and DSC analysis.

2.3. Characterization

2.3.1. Powder X-Ray Diffraction (PXRD) Analysis

The PXRD analysis was accomplished on PANalytical X'Pert Pro powder X-ray diffractometer system. The X-ray of wavelength 1.54056 Å was used. The data was collected in the form of a chart of the Bragg angle (2θ) vs. intensity, and a detailed table containing information on peak intensity counts, d value (Å), relative intensity (%), full width half maximum (FWHM) (θ°). From the PXRD results, the crystallite size (G) was calculated using X'Pert data collector and X'Pert high score plus processing software. The crystallite size (G) was calculated from the Scherrer equation [43, 44]. The method was based on the width of the diffraction patterns obtained in the X-ray reflected the crystalline region. The crystallite size (G) was calculated by using the following formula (1):

$$G = k\lambda / (b \cos\theta) \quad (1)$$

Where, k is the equipment constant (0.5), λ is the X-ray wavelength (0.154 nm); b in radians is the full-width at half of the peaks and θ the corresponding Bragg angle.

Percent change in crystallite size (G) was calculated using following equation (2):

$$\% \text{ change in crystallite size} = \frac{[G_{\text{Treated}} - G_{\text{Control}}]}{G_{\text{Control}}} \times 100 \quad (2)$$

Where, G_{Control} and G_{Treated} are the crystallite size of the control and Biofield Energy Treated samples, respectively.

A total of ~500 mg of the control and Biofield Energy Treated samples individually were used for the analysis and prepared by back loading technique using the sample preparation kit. The sample was spread on the holder ring in sufficient quantity to fill the ring cavity. It was then pressed down using powder press block and scrap the powder that was in surplus using a glass slide to get densely packed specimen. Consequently, the bottom plate was placed onto the holder ring and clamp in position. The sample holder was then removed from the sample preparation table by turning it upside down. A smooth surface of the sample was obtained to ensure optimum results.

2.3.2. Particle Size Distribution (PSD) Analysis

The average particle size and particle size distribution were analyzed using Malvern Mastersizer 2000, UK with a detection range between 0.01 μm to 3000 μm. The sample unit was filled with the dispersant medium (sunflower oil) and operated the stirrer at 2500 rpm. Alignment of the optics was done and the background measurement was taken. After the background measurement, the sample was added into the sample unit with constant monitoring the obscuration and stopped the addition of sample when the obscuration reached in between 15% to 20%. When the obscuration was stable, the measurement was taken twice and the average of two measurements was taken. The average histogram of the two measurements was recorded. The data was presented in the table format which includes particle size (μm). Also, the values at below 10% level (d_{10}), 50% level (d_{50}), and 90% level (d_{90}) were calculated from the histogram and the calculations such as surface area (m^2/g) were done by using software Mastersizer 2000.

Percent change in particle size (d) for at below 10% level (d_{10}), 50% level (d_{50}), and 90% level (d_{90}) was calculated using following equation (3):

$$\% \text{ change in particle size} = \frac{[d_{\text{Treated}} - d_{\text{Control}}]}{d_{\text{Control}}} \times 100 \quad (3)$$

Where, d_{Control} and d_{Treated} are the particle size (μm) for at below 10% level (d_{10}), 50% level (d_{50}), and 90% level (d_{90}) of the control and Biofield Energy Treated samples, respectively.

Percent change in surface area (S) was calculated using following equation (4):

$$\% \text{ change in surface area} = \frac{[S_{\text{Treated}} - S_{\text{Control}}]}{S_{\text{Control}}} \times 100 \quad (4)$$

Where, S_{Control} and S_{Treated} are the surface area of the control and Biofield Energy Treated samples, respectively.

2.3.3. Fourier Transform Infrared (FT-IR) Spectroscopy

FT-IR spectroscopy of zinc chloride was performed on Spectrum two (Perkin Elmer, USA) Fourier transform

infrared spectrometer with the frequency array of 400-4000 cm^{-1} by using pressed KBr disk technique.

2.3.4. Ultraviolet-Visible Spectroscopy (UV-Vis) Analysis

The UV-Vis spectral analysis was carried out using Shimadzu UV-2450 with UV Probe, Japan. The spectrum was recorded using 1 cm quartz cell that has a slit width of 1.0 nm. The wavelength range chosen for recording the spectra was 190-800 nm. The absorbance spectra (in the range of 0.2 to 0.9) and wavelength of maximum absorbance (λ_{max}) were recorded.

2.3.5. Differential Scanning Calorimetry (DSC)

The analysis was performed using the DSC Q20 (TA Instruments, USA) differential scanning calorimeter. A total of ~5 mg sample was weighed and sealed in aluminum pans and equilibrated at 25°C and heated up to 450°C at the

heating rate of 10°C/min under nitrogen gas as purge atmosphere with a flow rate of 50 mL/min. The value for onset, endset, peak temperature, peak height (mJ or mW), peak area, and change in heat (J/g) for each peak were recorded.

Percent change in the melting/decomposition temperature (T) was calculated using following equation (5):

$$\% \text{ change in melting/decomposition temperature} = \frac{[T_{\text{Treated}} - T_{\text{Control}}]}{T_{\text{Control}}} \times 100 \quad (5)$$

Where, T_{Control} and T_{Treated} are the decomposition temperature of the control and Biofield Energy Treated samples, respectively.

Percent change in the enthalpy of melting/decomposition (ΔH) was calculated using following equation (6):

$$\% \text{ change in enthalpy of melting/decomposition} = \frac{[\Delta H_{\text{Treated}} - \Delta H_{\text{Control}}]}{\Delta H_{\text{Control}}} \times 100 \quad (6)$$

Where, $\Delta H_{\text{Control}}$ and $\Delta H_{\text{Treated}}$ are the enthalpy of decomposition of the control and Biofield Energy Treated samples, respectively.

3. Results and Discussion

3.1. Powder X-ray Diffraction (PXRD) Analysis

The PXRD diffractograms of the control and Biofield Energy Treated zinc chloride are presented in Figure 1. Both the samples displayed very sharp and intense peaks in their respective diffractogram indicating the crystalline nature of zinc chloride. PXRD data such as Bragg angle (2θ), full width half maximum (FWHM) (θ°), relative intensity (%), and crystallite size (G) for the control and Biofield Energy Treated zinc chloride are presented in Table 1. The crystallite size was calculated with the help of Scherrer equation [43, 44]. The crystallite size of the control and Biofield Energy Treated samples were found to be similar at 2θ equal to nearly 29.3°, 49.2°, and 74.6° (Table 1, entry 3, 7, and 12).

The crystallite size of the Biofield Energy Treated zinc chloride at 2θ equal to nearly 25.4°, 34.9°, 38.2°, 52.3°, 56.2°, 58.4°, and 61.7° (Table 1, entry 2, 4, 5, and 8-11) was significantly increased in the range of 25.04% to 166.82% compared with the control sample. Consequently, at 2θ equal to 15.6°, 48.6°, and 91.9° (Table 1, entry 1, 6, and 13), the crystallite size of the Biofield Energy Treated zinc chloride was significantly decreased by 16.73%, 33.35%, and 19.95%, respectively compared with the control sample. However, the average crystallite size of the Biofield Energy Treated sample was significantly increased by 16.43% (Table 1, entry 14) compared to the control sample. The PXRD diffractograms of both the control and Biofield Energy Treated zinc chloride displayed greater intensity at Bragg's angle (2θ) equal to 48.6° (Table 1, entry 6). Beside this, the relative intensities of the PXRD peaks (Table 1) in the Biofield Energy Treated sample were significantly altered compared with the control sample.

Table 1. PXRD data of the control and Biofield Energy Treated zinc chloride.

Entry No.	Bragg angle (2θ)		Relative Intensity (%)		FWHM ($^\circ 2\theta$)		Crystallite size (G, nm)		
	Control	Treated	Control	Treated	Control	Treated	Control	Treated	% change*
1	15.56	15.60	39.59	41.02	0.0836	0.1004	53.14	44.25	-16.73
2	25.37	25.43	30.40	31.09	0.1673	0.1338	26.97	33.73	25.05
3	29.29	29.21	29.26	41.66	0.0502	0.0502	90.64	90.63	-0.02
4	34.90	34.88	91.00	38.44	0.0816	0.0502	56.56	91.93	62.54
5	38.22	38.21	15.20	20.55	0.2342	0.1338	19.90	34.83	75.03
6	48.55	48.60	100.00	100.00	0.0669	0.1004	72.21	48.13	-33.35
7	49.12	49.19	71.85	72.97	0.0612	0.0612	79.12	79.14	0.03
8	52.18	52.27	12.22	13.29	0.1632	0.1224	30.05	40.08	33.38
9	56.09	56.16	21.75	46.38	0.1020	0.0816	48.93	61.18	25.04
10	58.24	58.36	12.93	41.25	0.1632	0.0612	30.90	82.44	166.82
11	61.70	61.54	19.58	19.69	0.1020	0.0612	50.31	83.78	66.53
12	74.54	74.64	12.18	16.35	0.0816	0.0816	67.87	67.92	0.07
13	91.88	91.95	18.72	25.55	0.0816	0.1020	77.74	62.23	-19.95
14	Average crystallite size						54.18	63.10	16.43

FWHM: Full width half maximum, *denotes the percentage change in the crystallite size of the Biofield Energy Treated sample with respect to the control sample.

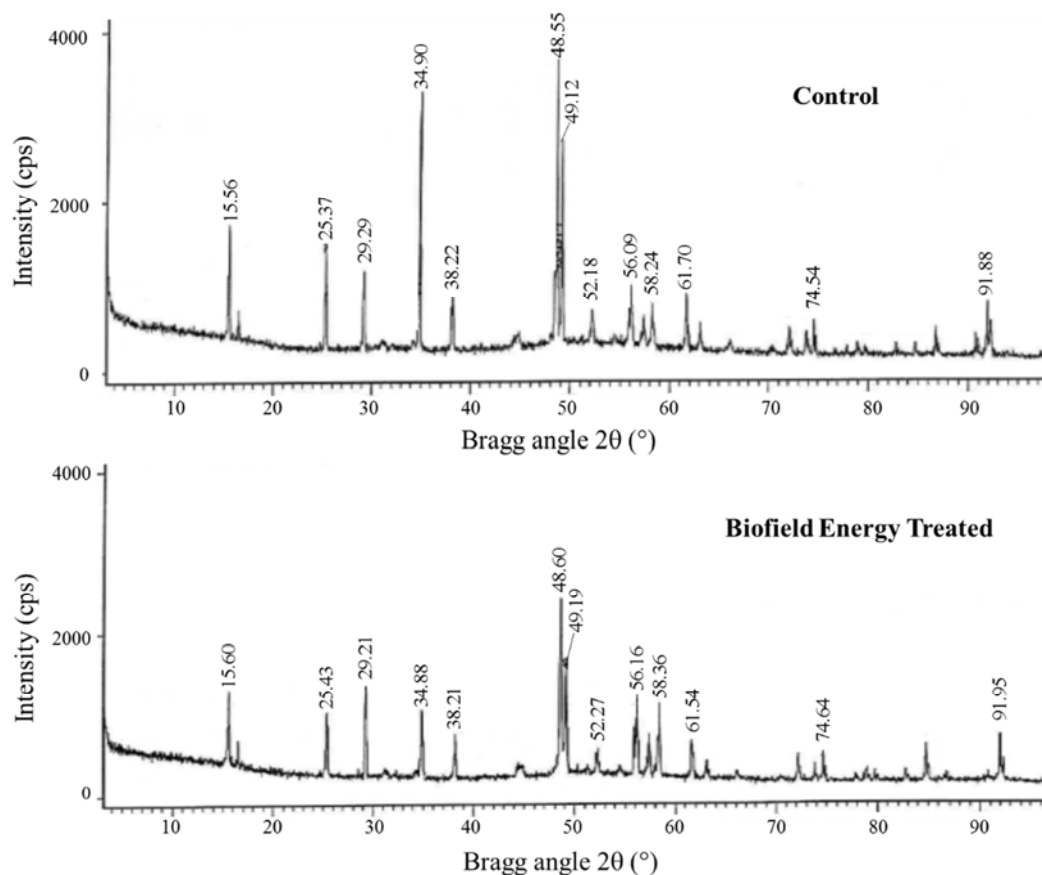


Figure 1. PXRD diffractograms of the control and Biofield Energy Treated zinc chloride.

The literature reported that the changes in the XRD patterns, such as crystallite size and relative intensities, indicated the modification of the morphology of the crystal as well as the proof of polymorphic transition [45-47]. As the crystal morphology of the Biofield Energy Treated sample was altered compared with the control sample, the treated sample might be a new polymorphic form of zinc chloride. The crystal pattern, size and even polymorphic form of a pharmaceutical play important roles in drug solubility, dissolution, and bioavailability. It has been reported in the literature that the alteration in crystal morphology has the significant impact on the *in vitro* rate of dissolution, which is related with the bioavailability of orally administered pharmaceutical/nutraceutical [41]. Thus, it can be concluded that the Biofield Energy Healing Treatment might be a very useful method for improving the bioavailability of zinc chloride.

3.2. Particle Size Distribution (PSD) Analysis

The particle size values and the surface area of the control and Biofield Energy Treated zinc chloride were investigated and the results are presented in Table 2. It was evident that the Biofield Energy Treatment has the ability to reduce the particle size of zinc chloride with respect to the control sample. The size of the particles in the Biofield Energy Treated sample at d_{10} was decreased by 2.42%, while at d_{50} , and d_{90} it was increased slightly by 0.43%, and 0.57%,

respectively compared to the control sample.

Table 2. Particle size data (d_{10} , d_{50} , and d_{90}) and surface area of the control and Biofield Energy Treated zinc chloride.

Parameter	d_{10} (μm)	d_{50} (μm)	d_{90} (μm)	Surface area (m^2/g)
Control	0.992	2.573	5.975	3.07
Biofield Energy Treated	0.968	2.584	6.009	3.12
Percent change (%)*	-2.42	0.43	0.57	1.63

*denotes the percentage change in the particle size data (d_{10} , d_5 , and d_{90}) and surface area of the Biofield Energy Treated sample with respect to the control sample.

The particle size and surface area have the direct effect on the dissolution of a compound in a solvent, because the surface energy influenced by the surface area and chemical affinity is the driving factor for dissolution efficiency [48, 49]. The surface area for both the control and Biofield Energy Treated zinc chloride samples were analyzed and the results are presented in Table 2. The surface area of the Biofield Energy Treated sample ($3.12 \text{ m}^2/\text{g}$) was increased by 1.63% compared with the control sample ($3.07 \text{ m}^2/\text{g}$).

Poorly crystallized compounds possess more surface area and higher exchange capacities than well-crystallized compounds [50]. In addition, the variation of the crystal morphology in the Biofield Energy Treated sample, which was well-supported from PXRD data, may cause to alter the surface area of the Biofield Energy Treated zinc chloride

compared with the control sample. It has been well established that the particle size, shape and surface area of pharmaceutical compounds have an important impact on solubility, dissolution and *in vivo* bioavailability, as well as supports the designing of new drug delivery systems [51, 52]. Reducing the particle size and higher surface area would enhance the solubility of the solid particles, and consequently increase the dissolution rate and bioavailability [53]. Thus, it is assumed that the Biofield Energy Treated zinc chloride might be more soluble, absorbed at a faster rate and may possibly have more bioavailability than the untreated zinc chloride.

3.3. Fourier Transform Infrared (FT-IR) Spectroscopy

The FT-IR spectra of the control and Biofield Energy Treated samples of zinc chloride are presented in Figure 2. The characteristic peaks were observed at 3908, 3589, 3518, 1930, 1607, 980, and 511 cm^{-1} in the control FT-IR spectrum (Figure 2). On the other hand, the FT-IR spectrum of the

Biofield Energy Treated zinc chloride showed the notable absorption bands at 3902, 3589, 3519, 1931, 1607, 988, and 508 cm^{-1} (Figure 2). The strong and broad absorption band centered near 3589 cm^{-1} and the H-O-H bending motion at 1607 cm^{-1} that were due to the presence of the lattice water was found in the FT-IR spectrum of both the control and Biofield Energy Treated samples (Figure 2). Literature suggested that the metal-halide stretching absorption band for inorganic materials generally found in the 750-100 cm^{-1} region [54]. In control spectrum, metal-halogen (Zn-Cl) stretching was found at 511 cm^{-1} , whereas it was shifted downward in the Biofield Energy Treated sample at 508 cm^{-1} . Overall, FT-IR results suggested that there might be a significant impact of Biofield Energy Treatment on zinc chloride at the atomic level to decrease the force constant of Zn-Cl bonds. This result also supported with the PXRD data that Biofield Energy Treated sample might be the new polymorphic form of zinc chloride.

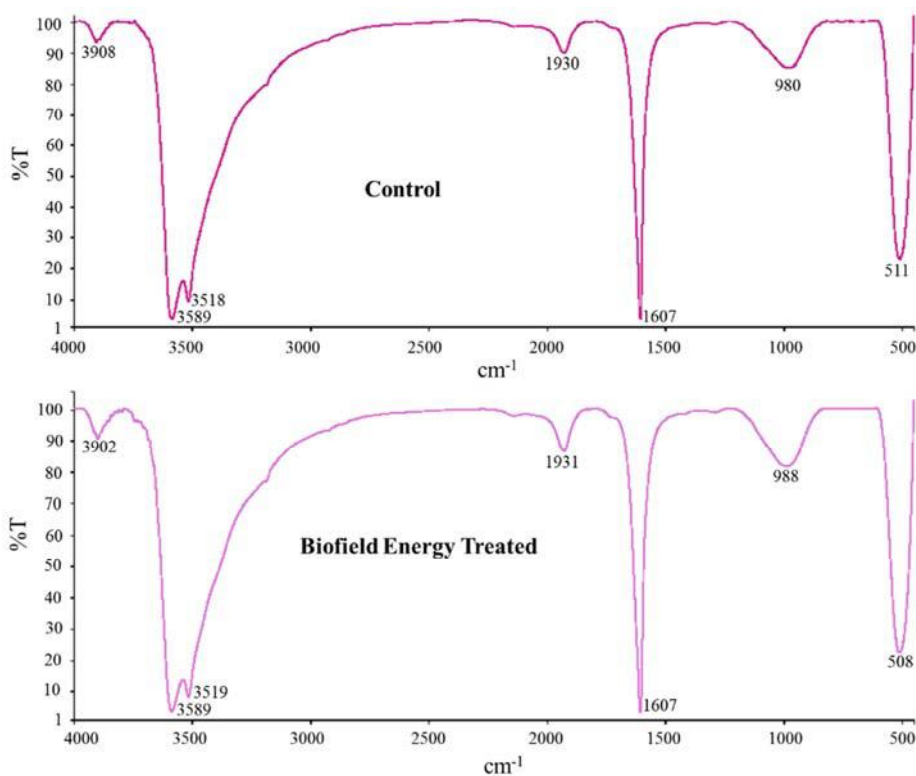


Figure 2. FT-IR spectra of the control and Biofield Energy Treated zinc chloride.

3.4. Ultraviolet-Visible Spectroscopy (UV-Vis) Analysis

The UV-visible spectra of both the control and Biofield Energy Treated zinc chloride are presented in Figure 3. The spectra of the control and treated sample showed the maximum absorbance (λ_{max}) at 196.4 nm and 196.2 nm, respectively, along with the minor shift of absorbance maxima from 1.9084 (control sample) to 1.9259 (Biofield Energy Treated sample). The UV absorbance occurs due to the diverse type of energy transitions from the singlet excited state such as

$\sigma \rightarrow \sigma^*$, $n \rightarrow \pi^*$, and $\pi \rightarrow \pi^*$. These type of electronic transitions occur when the difference in energy between the lowest unoccupied molecular orbital (LUMO) and the highest occupied molecular orbital (HOMO) is significantly higher than the activation energy of the compound [55]. The UV-vis spectroscopic analysis of zinc chloride displayed no change in the λ_{max} of the Biofield Energy Treated sample compared to the control sample. Thus, the structural configuration or activation energy of the Biofield Energy Treated sample was not changed compared with the control sample.

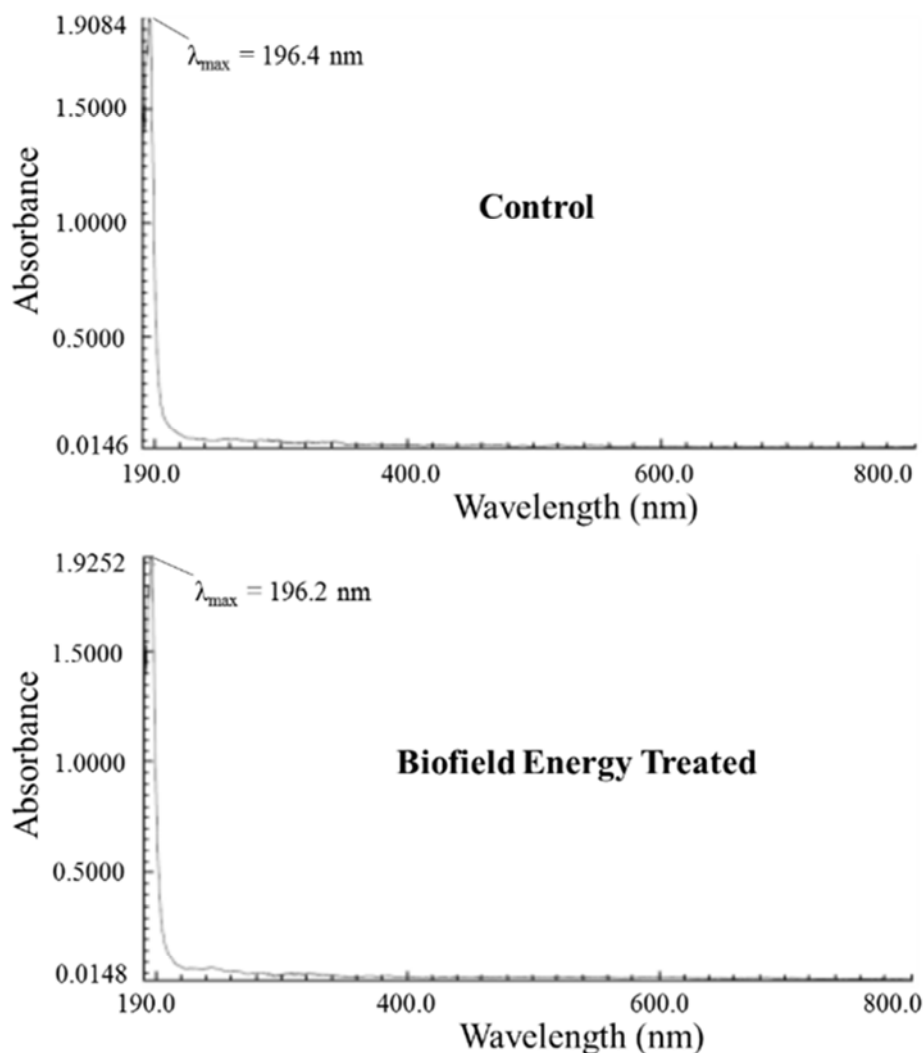


Figure 3. UV-vis spectra of the control and Biofield Energy Treated zinc chloride.

3.5. Differential Scanning Calorimetry (DSC) Analysis

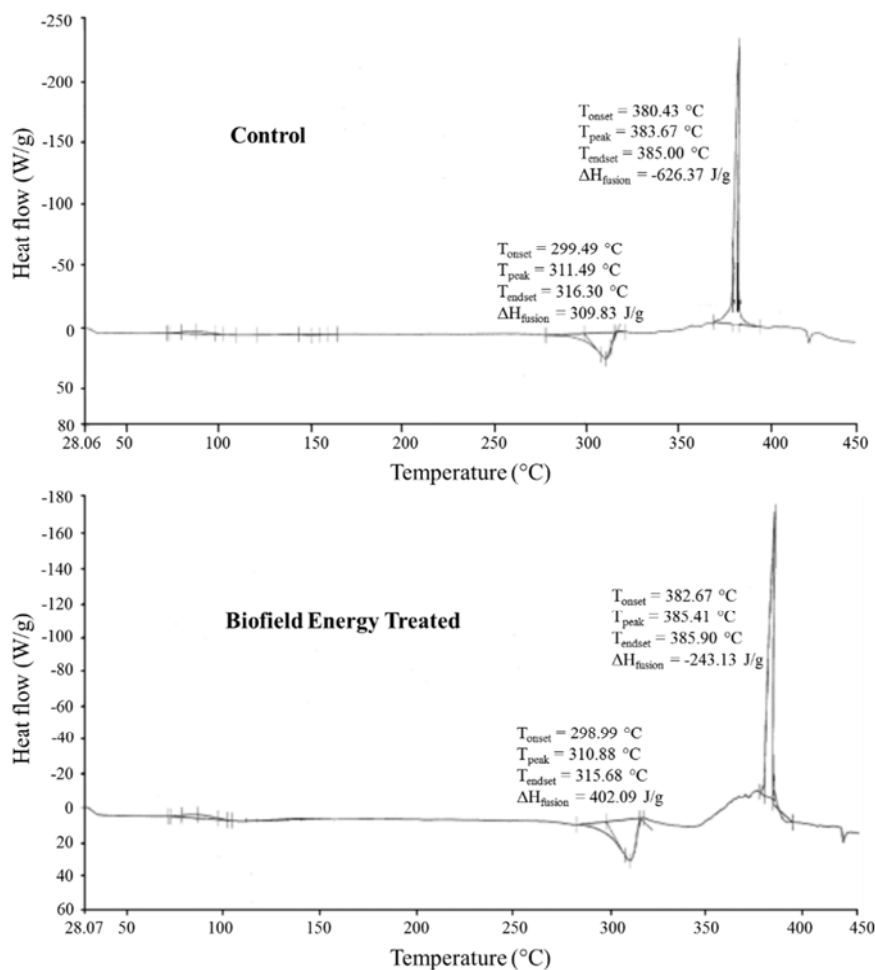
The DSC thermograms of the control and Biofield Energy Treated zinc chloride are presented in Figure 4. The melting/decomposition temperature and enthalpy of fusion/decomposition of the control and treated zinc chloride are presented in Table 3. The melting temperature was slightly decreased by 0.20% in the Biofield Energy Treated zinc chloride (310.88°C) compared with the control sample (311.49°C). The onset and the endset melting temperature were also decreased slightly by 0.17% and 0.20%, respectively in the Biofield Energy Treated sample compared with the control sample. The latent heat of fusion or enthalpy of fusion (ΔH_{fusion}) of the control and Biofield Energy Treated zinc chloride was 309.83 J/g and 402.09 J/g, respectively. The ΔH_{fusion} of the Biofield Energy Treated sample was significantly increased by 29.78% compared with the control sample. Similarly, the decomposition temperature of the Biofield Energy Treated zinc chloride (385.41°C) was increased by 0.45% compared to the control sample

(383.67°C). The onset and endset decomposition temperature were also increased slightly by 0.59% and 0.23%, respectively in the Biofield Energy Treated sample compared with the control sample. The enthalpy of decomposition ($\Delta H_{\text{decomposition}}$) of the control and Biofield Energy Treated zinc chloride were 243.13 J/g and 626.37 J/g, respectively. The $\Delta H_{\text{decomposition}}$ of the Biofield Energy Treated sample was significantly decreased by 61.18% compared with the control sample. Overall, the results indicated that the Biofield Energy Treated zinc chloride needs more energy in the form of latent heat to undergo the process of melting after Biofield Energy Treatment. It is anticipated that Biofield Energy Treatment might enhance the internal energy in zinc chloride that leads to increase of the decomposition temperature. Thus, the thermal stability of the Biofield Energy Treated sample might be more compared with the control sample. The enhanced thermal stability of a drug is an advantage in the maintenance of the product in long term storage condition throughout the shelf-life for any pharmaceutical formulations [56].

Table 3. The latent heat of fusion (J/g), enthalpy of decomposition (J/g) and melting/decomposition temperature (°C) values of the control and Biofield Energy Treated zinc chloride.

Endothermic peak	Description	T _{onset} (°C)	T _{peak} (°C)	T _{endset} (°C)	ΔH (J/g)
Endothermic peak	Control sample	299.49	311.49	316.30	309.83
	Biofield Energy Treated sample	298.99	310.88	315.68	402.09
% Change*		-0.17	-0.20	-0.20	29.78
Exothermic peak	Control sample	380.43	383.67	385.00	626.37
	Biofield Energy Treated sample	382.67	385.41	385.90	243.13
% Change*		0.59	0.45	0.23	-61.18

T_{onset}: Onset melting/decomposition temperature, T_{peak}: Peak melting/decomposition temperature, T_{endset}: Endset melting/decomposition temperature, ΔH: Enthalpy of fusion/decomposition, *denotes the percentage change of the Biofield Energy Treated sample with respect to the control sample.

**Figure 4.** DSC thermograms of the control and Biofield Energy Treated zinc chloride.

4. Conclusions

The current experimental results revealed that The Trivedi Effect[®] - Consciousness Energy Healing Treatment had the significant effects on the physical, structural, thermal, and behavioral properties of zinc chloride. The PXRD data displayed the significant alteration of the crystallite size and relative intensities of the PXRD peaks in The Trivedi Effect[®] treated sample compared with the control sample. The average crystallite size of the treated sample was significantly increased by 16.43% compared with the control sample. The surface area of the treated sample was increased by 1.63% compared with the control sample. The FT-IR

spectroscopic analysis revealed that Zn-Cl stretching in the control sample was 511 cm⁻¹, whereas it was significantly shifted downward to 508 cm⁻¹ in the treated sample. The DSC analysis exhibited that the melting temperature was not altered significantly, while decomposition temperature was increased by 0.45% in the treated sample compared to the control sample. The latent heat of fusion of the treated sample was significantly increased by 29.78% compared with the control sample. Similarly, the enthalpy of decomposition of the treated sample was significantly altered by 61.18% compared with the control sample. Thus, the Biofield Energy Treatment improve the thermal stability of the treated sample compared to the control sample. Briefly, The Trivedi Effect[®]

- Consciousness Energy Healing Treatment might introduce a new thermally stable polymorphic form of zinc chloride that could be more soluble and bioavailable compared with the untreated compound. Hence, Biofield Energy Treated zinc chloride would be very useful to design better nutraceutical and/or pharmaceutical formulations that might offer better therapeutic response against immunological disorders, inflammatory diseases, aging, stress, Wilson's disease, cancer, viral diseases, parakeratosis, anorexia, hypogeusia, geophagia, dysosmia, hypogonadism, growth retardation, etc.

Abbreviations

DSC: Differential scanning calorimetry, FT-IR: Fourier transform infrared spectroscopy, FWHM: Full width half maximum, G: Crystallite size, HOMO: Highest energy occupied molecular orbital, LUMO: Lowest energy unoccupied molecular orbital, T_{onset} : Onset melting/decomposition temperature, T_{peak} : Peak melting/decomposition temperature, T_{endset} : Endset melting/decomposition temperature, ΔH : Enthalpy of fusion/decomposition, UV-vis: Ultraviolet-visible spectroscopy, PSD: Particle size distribution, PXRD: Powder X-ray diffraction.

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References

- [1] Ronconi L, Sadler PJ (2008) Applications of heteronuclear NMR spectroscopy in biological and medicinal inorganic chemistry. *Coordn Chem Rev* 252: 2239-2277.
- [2] Berg JM, Shi Y (1996) The galvanization of biology: A growing appreciation for the roles of zinc. *Science* 271: 1081-1085.
- [3] Higdon JV, Ho E (2005) In: M. Gielen, E. R. T. Tiekink (Eds.), *Metallotherapeutic drugs and metal-based diagnostic agents: The use of metals in medicine*, Wiley-VCH, Weinheim, p. 237.
- [4] Brewer GJ (2001) Zinc acetate for the treatment of Wilson's disease. *Expert Opin Pharmacother* 2: 1473-1477.
- [5] Prasad AS (1979) Clinical, biochemical, and pharmacological role of zinc. *Ann Rev Pharmacol Toxicol* 19: 393-426.
- [6] Supuran CT (2008) Carbonic anhydrases: Novel therapeutic applications for inhibitors and activators. *Nat Rev Drug Dis* 7: 168-181.
- [7] Elmes ME (1975) Letter: Zinc in human medicine. *Lancet* 2: 549.
- [8] Mazumder PM, Pattnayak S, Parvani H, Sasmal D, Rathinavelusamy P (2012) Evaluation of immunomodulatory activity of *Glycyrrhiza glabra* L. roots in combination with zing. *Asian Pac J Trop Biomed* 2: S15-S20.
- [9] Brynstad J, Yakel HL (1978) Preparation and structure of anhydrous zinc chloride. *Inorg Chem* 17: 1376-1377.
- [10] Kasture AV, Wadodkar SG (2008) A text book of pharmaceutical chemistry-1, Nirali Prakashan, 25th Ed., Pune, India.
- [11] Mahadik KR, Kuchekar BS (2008) Concise inorganic pharmaceutical chemistry, Nirali Prakashan, 25th Ed., Pune, India.
- [12] McDaniel S, Goldman GD (2002) Consequences of using escharotic agents as primary treatment for nonmelanoma skin cancer. *Arch Dermatol* 138: 1593-1596.
- [13] Hu J, Yang Z, Wang J, Yu J, Guo J, Liu S, Qian C, Song L, Wu Y, Cheng J (2016) Zinc chloride transiently maintains mouse embryonic stem cell pluripotency by activating Stat3 signaling. *PLoS One* 11: e0148994.
- [14] Fukuyama Y, Kawarai S, Tezuka T, Kawabata A, Maruo T (2016) The palliative efficacy of modified Mohs paste for controlling canine and feline malignant skin wounds. *Vet Q* 1: 1-7.
- [15] Yakimovskii AF, Kryzhanovskaya SY (2015) Zinc chloride and zinc acetate injected into the neostriatum produce opposite effect on locomotor behavior of rats. *Bull Exp Biol Med* 160: 281-282.
- [16] Stenger VJ (1999) Bioenergetic fields. *Sci Rev Alternative Med* 3: 16-21.
- [17] Rogers, M (1989) "Nursing: A Science of Unitary Human Beings." In J. P. Riehl-Sisca (ed.) *Conceptual Models for Nursing Practice*. 3rd Edn. Norwalk: Appleton & Lange.
- [18] Rosa L, Rosa E, Sarner L, Barrett S (1998) A close look at therapeutic touch. *Journal of the American Medical Association* 279: 1005-1010.
- [19] Warber SL, Cornelio D, Straughn, J, Kile G (2004) Biofield energy healing from the inside. *J Altern Complement Med* 10: 1107-1113.
- [20] Koithan M (2009) Introducing complementary and alternative therapies. *J Nurse Pract* 5: 18-20.
- [21] Trivedi MK, Patil S, Shettigar H, Mondal SC, Jana S (2015) The potential impact of biofield treatment on human brain tumor cells: A time-lapse video microscopy. *J Integr Oncol* 4: 141.
- [22] Trivedi MK, Patil S, Shettigar H, Bairwa K, Jana S (2015) Phenotypic and biotypic characterization of *Klebsiella oxytoca*: An impact of biofield treatment. *J Microb Biochem Technol* 7: 203-206.
- [23] Trivedi MK, Patil S, Shettigar H, Gangwar M, Jana S (2015) An effect of biofield treatment on multidrug-resistant *Burkholderia cepacia*: A multihost pathogen. *J Trop Dis* 3: 167.
- [24] Trivedi MK, Patil S, Shettigar H, Gangwar M, Jana S (2015) An evaluation of biofield treatment on susceptibility pattern of multidrug resistant *Stenotrophomonas maltophilia*: An emerging global opportunistic pathogen. *Clin Microbiol* 4: 211.
- [25] Trivedi MK, Patil S, Shettigar H, Gangwar M, Jana S (2015) *In vitro* evaluation of biofield treatment on cancer biomarkers involved in endometrial and prostate cancer cell lines. *J Cancer Sci Ther* 7: 253-257.

- [26] Trivedi MK, Branton A, Trivedi D, Nayak G, Mondal SC, Jana S (2015) Evaluation of plant growth regulator, immunity and DNA fingerprinting of biofield energy treated mustard seeds (*Brassica juncea*). *Agriculture, Forestry and Fisheries* 4: 269-274.
- [27] Trivedi MK, Branton A, Trivedi D, Nayak G, Mondal SC, Jana S (2015) Biochemical differentiation and molecular characterization of biofield treated *Vibrio parahaemolyticus*. *American Journal of Clinical and Experimental Medicine* 3: 260-267.
- [28] Trivedi MK, Branton A, Trivedi D, Gangwar M, Jana S (2015) Antimicrobial susceptibility, biochemical characterization and molecular typing of biofield treated *Klebsiella pneumoniae*. *J Health Med Inform* 6: 206.
- [29] Trivedi MK, Tallapragada RM, Branton A, Trivedi D, Nayak G, Latiyal O, Jana S (2015) The potential impact of biofield energy treatment on the physical and thermal properties of silver oxide powder. *International Journal of Biomedical Science and Engineering* 3: 62-68.
- [30] Trivedi MK, Tallapragada RM, Branton A, Trivedi D, Nayak G, Latiyal O, Jana S (2015) Analysis of physical, thermal, and structural properties of biofield energy treated molybdenum dioxide. *International Journal of Materials Science and Applications* 4: 354-359.
- [31] Trivedi MK, Branton A, Trivedi D, Nayak G, Bairwa K, Jana S (2015) Spectroscopic characterization of disulfiram and nicotinic acid after biofield treatment. *J Anal Bioanal Tech* 6: 265.
- [32] Trivedi MK, Branton A, Trivedi D, Shettigar H, Bairwa K, Jana S (2015) Fourier transform infrared and ultraviolet-visible spectroscopic characterization of biofield treated salicylic acid and sparfloxacin. *Nat Prod Chem Res* 3: 186.
- [33] Trivedi MK, Tallapragada RM, Branton A, Trivedi D, Nayak G, Mishra RK, Jana S (2015) Biofield treatment: A potential strategy for modification of physical and thermal properties of gluten hydrolysate and ipomoea macroelements. *J Nutr Food Sci* 5: 414.
- [34] Trivedi MK, Nayak G, Patil S, Tallapragada RM, Jana S, Mishra RK (2015) Bio-field Treatment: An effective strategy to improve the quality of beef extract and meat infusion powder. *J Nutr Food Sci* 5: 389.
- [35] Trivedi MK, Branton A, Trivedi D, Nayak G, Latiyal O, Jana S (2015) Evaluation of biofield treatment on atomic and thermal properties of ethanol. *Organic Chem Curr Res* 4: 145.
- [36] Trivedi MK, Branton A, Trivedi D, Nayak G, Singh R, Jana S (2015) Physical, thermal and spectroscopic studies on biofield treated *p*-dichlorobenzene. *Biochem Anal Biochem* 4: 204.
- [37] Trivedi MK, Branton A, Trivedi D, Nayak G, Bairwa K, Jana S (2015) Physicochemical and spectroscopic characterization of biofield energy treated *p*-anisidine. *Pharm Anal Chem Open Access* 6: 102.
- [38] Trivedi MK, Branton A, Trivedi D, Nayak G, Gangwar M, Jana S (2015) Agronomic characteristics, growth analysis, and yield response of biofield treated mustard, cowpea, horse gram, and groundnuts. *International Journal of Genetics and Genomics* 3: 74-80.
- [39] Trivedi MK, Branton A, Trivedi D, Nayak G, Gangwar M, Jana S (2015) Analysis of genetic diversity using simple sequence repeat (SSR) markers and growth regulator response in biofield treated cotton (*Gossypium hirsutum* L.). *American Journal of Agriculture and Forestry* 3: 216-221.
- [40] Chereson R (2009) Bioavailability, bioequivalence, and drug selection. In: Makoid CM, Vuchetich PJ, Banakar UV (Eds) *Basic pharmacokinetics* (1st Edn) Pharmaceutical Press, London.
- [41] Blagden N, de Matas M, Gavan PT, York P (2007) Crystal engineering of active pharmaceutical ingredients to improve solubility and dissolution rates. *Adv Drug Deliv Rev* 59: 617-630.
- [42] Trivedi MK, Mohan TRR (2016) Biofield energy signals, energy transmission and neutrinos. *American Journal of Modern Physics* 5: 172-176.
- [43] Alexander L, Klug HP (1950) Determination of crystallite size with the X-Ray Spectrometer. *J App Phys* 21: 137.
- [44] Langford JI, Wilson AJC (1978) Scherrer after sixty years: A survey and some new results in the determination of crystallite size. *J Appl Cryst* 11: 102-113.
- [45] Raza K, Kumar P, Ratan S, Malik R, Arora S (2014) Polymorphism: The phenomenon affecting the performance of drugs. *SOJ Pharm Pharm Sci* 1: 10.
- [46] Brittain HG (2009) *Polymorphism in pharmaceutical solids in Drugs and Pharmaceutical Sciences*, volume 192, 2nd Edn, Informa Healthcare USA, Inc., New York.
- [47] Thiruvengadam E, Vellaisamy G (2014) Polymorphism in pharmaceutical ingredients a review. *World Journal of Pharmacy and Pharmaceutical Sciences* 3: 621-633.
- [48] <http://www.dissolution.com/ddg/showthread.php?2366-Surface-Area-vs-Particle-Size>.
- [49] Mosharrof M, Nyström C (1995) The effect of particle size and shape on the surface specific dissolution rate of micro-sized practically insoluble drugs. *Int J Pharm* 122: 35-47.
- [50] Murray HH, Lyons SC (1960) Further correlation of kaolinite crystallinity with chemical and physical properties. *Clays Clay Miner* 8: 11-17.
- [51] Sun J, Wang F, Sui Y, She Z, Zhai W, Wang C, Deng Y (2012) Effect of particle size on solubility, dissolution rate, and oral bioavailability: Evaluation using coenzyme Q₁₀ as naked nanocrystals. *Int J Nanomed* 7: 5733-5744.
- [52] Khadka P, Ro J, Kim H, Kim I, Kim JT, Kim H, Cho JM, Yun G, Lee J (2014) Pharmaceutical particle technologies: An approach to improve drug solubility, dissolution and bioavailability. *Asian J Pharm Sci* 9: 304-316.
- [53] Buckton G, Beezer AE (1992) The relationship between particle size and solubility. *Int J Pharmaceutics* 82: R7-R10.
- [54] Stuart BH (2004) *Infrared spectroscopy: Fundamentals and applications in Analytical Techniques in the Sciences*. John Wiley & Sons Ltd., Chichester, UK.
- [55] Hesse M, Meier H, Zeeh B (1997) *Spectroscopic methods in organic chemistry*, Georg Thieme Verlag Stuttgart, New York.
- [56] Bajaj S, Singla D, Sakhuja N (2012) Stability testing of pharmaceutical products. *J App Pharm Sci* 2: 129-138.