

## RESEARCH ARTICLE

# Flakes Size-Dependent Optical and Electrochemical Properties of MoS<sub>2</sub>

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**Abstract: Background:** Molybdenum disulfide (MoS<sub>2</sub>) is a transition metal dichalcogenides and has some interesting and promising properties. MoS<sub>2</sub> has direct and indirect band gaps depending on its crystalline structure. In addition, its sheets morphology makes it a good candidate for supercapacitor applications.

**Objective:** The aim of this work is to study the effect of MoS<sub>2</sub> flakes size on its optical and electrochemical properties.

**Method:** MoS<sub>2</sub> with different flakes sizes were prepared by exfoliation method. The exfoliation was performed by sonication of MoS<sub>2</sub> powder in N,N-Dimethylformamide followed by different centrifugation speeds. UV-Vis spectra illustrated the optical energy gap was inversely proportional to the MoS<sub>2</sub> flakes size.

**Results:** Absorption coefficient values indicated that the exfoliation reduced the number of layers. Symmetric supercapacitor was made from two MoS<sub>2</sub> electrodes and tested in 6 M KOH electrolyte. The specific capacitance was found to be dramatically increased with decreasing flakes size (9.5 and 4.5 mF/cm<sup>2</sup> for 0.26 and 0.98 μm flakes size, respectively).

**Conclusion:** These findings recommend that MoS<sub>2</sub> can be the excellent electrode material for supercapacitor.

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## 1. INTRODUCTION

Molybdenum disulfide (MoS<sub>2</sub>) nowadays has great attention in research fields due to its exciting properties. MoS<sub>2</sub> is a transition metal dichalcogenides (TMDs) [1] in which its properties are highly dependent on the crystalline structure and sizes to determine the metallic (1T) or semiconductor (2H) phase [2]. Moreover, MoS<sub>2</sub> shows a direct band gap in monolayer 2H-MoS<sub>2</sub> and indirect band gap in bulk 2H-MoS<sub>2</sub> [3]. The tunable band gap in MoS<sub>2</sub> allows it to be used in electronic and optical applications [4-6]. Apart from the electronic application, it is used as a lubricant due to its low friction coefficient and high lubricity and stability at high temperature [7]. Other unique properties of MoS<sub>2</sub> are high luminescence quantum efficiency [8], high exhibiting photoluminescence [9] and potential substitution

of the zero band gap materials [10]. The layered structure in MoS<sub>2</sub> which is held by weak van der Waals forces at the interlayer allows ions diffusion without significant any volume expansion. Because of this, MoS<sub>2</sub> was studied as a cathode material for batteries [11], supercapacitors, solar cells and conductors [12-15]. For the electrochemical applications, MoS<sub>2</sub> in single phase or composites in different morphologies are extensively studied. The 2D layer of MoS<sub>2</sub> showed a specific capacitance of 0.5 mF/cm<sup>2</sup> at 10 mV/s while the bulk exfoliated MoS<sub>2</sub> sheets showed specific capacitance of 2 mF/cm<sup>2</sup> at 10 mV/s [16]. The findings proved the bulk exfoliated MoS<sub>2</sub> sheets possess higher charge storage properties and render it to be the potential electrode material for supercapacitors. Various approaches were developed to exfoliate bulk MoS<sub>2</sub> crystal. Similar as the first report on graphene exfoliation, mechanical exfoliation was used to produce MoS<sub>2</sub> nanosheet [17-19]. In addition, the MoS<sub>2</sub> nanosheet also can be produced by the liquid phase exfoliation process by sonicating bulk MoS<sub>2</sub> in the organic solvent [16, 20]. The process of exfoliation produces MoS<sub>2</sub> nanosheets in various sizes, which behave differently especially in the context of electrochemistry as most of the

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