PAPER • OPEN ACCESS

Discover of GWLI as chemical flooding using SIT: experiment and analysis on key influence factor for oil recovery improvement

To cite this article: M Naser et al 2018 IOP Conf. Ser.: Earth Environ. Sci. 212 012072

View the article online for updates and enhancements.



You may also like

- Potential impacts of 1.5 °C and 2 °C global warming on rainfall onset, cessation and length of rainy season in West Africa Naomi Kumi and Babatunde J Abiodun
- <u>Projected effects of 1.5 °C and 2 °C global</u> warming levels on the intra-seasonal rainfall characteristics over the Greater <u>Horn of Africa</u> Masilin Gudoshava, Herbert O Misiani, Zewdu T Segele et al.
- <u>Risks of precipitation extremes over</u> <u>Southeast Asia: does 1.5 °C or 2 °C global</u> <u>warming make a difference?</u>
 Fei Ge, Shoupeng Zhu, Ting Peng et al.



This content was downloaded from IP address 196.159.17.223 on 27/11/2024 at 12:34

IOP Publishing

Discover of GWLI as chemical flooding using SIT: experiment and analysis on key influence factor for oil recovery improvement

M Naser¹, M Erhayem², A Hegaig³, M Abobakr¹, B Abobakr¹, and A Masood¹

¹ (Petroleum Engineering of Faculty of Mining and Energy, Sebha University, Libya)

² (Chemistry Engineering, Faculty of Science, Sebha University, Libya)

³ (Mechanical Engineering, Faculty of Engineering, Misurata University, Libya)

*corresponding author: madinaser2004@gmail.com

Abstract. Spontaneous Imbibition Test (SIT) is regarded as an important mechanism of oil recovery by water-flood, particularly in heterogeneous or fractured reservoirs where direct displacement of oil by water is usually poor. In this paper a new chemical technology (i.e. Gaberoun Water Lake Injection (GWLI)) has been discovered. It has several advantages which are relatively cheap and reliable. It potentially would have a wide range of applications in brine injection such as altering the rock wettability and responsible for increased ultimate recovery. Two stages of GWLI were discussed with changing pH and salinity. The first stage was used the core samples from carbonate and sandstone, which aged in oil for long time period. The second stage was used the core samples, which aged in oil for short time period. SIT was performed at 27, 30, 40, 50, 60 and 70°C with sandstone and carbonate rock with different aging time. The result showed that the oil recovery was decreased with a decrease in pH values and increased with a decrease salinity. However, the oil recovery in the long aging process was low comparing to the short aging process. The findings in this research that GWLI can be used for oil recovery processes.

Keywords: Gaberoun Water Lake Injection; Salinity; pH; Carbonate; Sandstone; Oil recovery, Aging time; Temperature; Spontaneous imbibition test.

1. Introduction

Improving oil recovery is recognized as the major target and challenge at the different stages of an oil field development. Among several methods in oil recovery, Injection of water into the reservoir is the usual the way to push the oil in front of the water towards the production well ([2]; [8]). Much oil remains in carbonate and sandstone oil reservoirs after water-flooding and in some cases in paleotransition zones, which result from the oil/water contact moving upward before discovery [13]. At this point, a high remaining oil saturation is left in reservoir, mainly because of wettability conditions, fractures, layers with large permeability contrasts, impermeable layers during imbibition. Capillary imbibition is described as a spontaneous penetration of a wetting phase into a porous media while displacing a non-wetting phase by means of capillary pressure, e.g., Water imbibes into an oilsaturated rock. It has been stated that the rate of imbibition increased with an increase in temperature due to reduction of oil-water interfacial tension, oil viscosity and water viscosity [[9]. It is clear that the crude oil-brine-rock interactions are responsible for the dramatic increase in oil recovery with

Content from this work may be used under the terms of the Creative Commons Attribution 3.0 licence. Any further distribution of this work must maintain attribution to the author(s) and the title of the work, journal citation and DOI. Published under licence by IOP Publishing Ltd 1

ICEMINE	IOP Publishing
IOP Conf. Series: Earth and Environmental Science 212 (2018) 012072	doi:10.1088/1755-1315/212/1/012072

temperature increase rather than changes related to the rock properties alone [11]. Another study is performed by using a similar chalk core sample and the results were comparable. Changes in temperature, using refined oil are not verified in this study [7].

The effect of temperature on oil recovery is the oil production rate increase or decrease with temperature, the studies shows that the temperatures has an important factor in increasing the cumulative oil recovery ([14]; [15]). Therefore the decrease in the viscosity ratio of oil and water due to increasing temperature result in oil being displaced more easily and the ultimate recovery being improved [3]. The time factor for equilibration has long been considered in restoring the original wetting condition of reservoir core samples. The length of time required to incubate the core samples, however, varies from one laboratory to another, ranging from a few days to months ([4]; [10]; [5]; [6]; [12]). A number of research works have been published in effect of hardness on oil recovery indicating that calcium ion Ca^{2+} , magnesium ion Mg^{2+} , and sulfate ion SO^{4}_{2-} in brine injection process. Both the concentration of Ca^{2+} and SO_4^{2-} at the carbonate surface increases as the temperature is increased. As Mg^{2+} is even able to displace Ca^{2+} from the carbonate rock at high temperature, it should also be able to displace the Ca²⁺ carboxylate complex from the surface. Investigation of effect of brine concentration on oil recovery often showed a significant increase in laboratory water-flood recoveries with a decrease in salinity for duplicate outcrop core plugs [16]. In this study was done to study the effect of pH and salinity on oil recovery by GWLI. The objective of this study to study the effect of GWLI on the oil recovery on sandstone and carbonate reservoirs at different temperature. To study the effect of change pH and the salinity on oil recovery.

2. Materials and Equipment

Figure 1 and 2 shows 12 cores of sandstone and 12 cores of carbonate from wells in the south of Libya were used. Oil sample used in this study with density of 0.764123 g/c.c and GWLI from Gaberoun Lake in Awbari desert. Ethylene Diamine Tetra Acetic Acid, E.D.T.A, was used to determine amount of calcium and magnesium in GWLI. pH=10 organizer solution was used determine amount of calcium and magnesium in GWLI. EBT Guider powder and Murexide Guider Powder were used as indicator to determine amount of calcium in GWLI. Sodium hydroxide solution, sodium chloride, toluene, barium chloride, conditioning agent was used in determine of sulfate concentration in GWLI.





Figure 1. Carbonate cores that used In this study

Figure 2. Sandstone cores that used in this study.

SIT was conducted in test tubes as shown in Figure 3. Soxhcelete Extractor Device was used in the process of cleaning core sample from oil. A pH meter, Spectrometer Device, Flame Photometer Device and Burette Bearer is used to measurement of properties of GWLI. Figure 4 shows Vacuum Chamber is used to saturate cores with distillate water and oil.

3. Methodology

The following three steps are used in the procedure for preparing and doing experiment. They are shown in Figure 5 and described in detail as follows.

1. GWLI analysis and preparation

- 2. Core sample preparation
- 3. Spontaneous imbibition test
 - Put the saturated core sample with oil in spontaneous imbibition pipes.
 - Fill spontaneous imbibition pipes that consist core sample by GWLI as showing in figure 8.
 - Close the pipe orifice by using aluminum foil and paper tape to prevent the oil from evaporation and left at room temperature until equilibrium is reached.
 - After the equilibrium is reached, the volume of the displacing oil is measured.
 - Then, raise the oven temperature to 30°C to displacing oil out of the sample until equilibrium is reached, the volume of the displacing oil is measured.
 - Repeat the PIT with raiseing the temperature to 40°C, 50°C, 60°C, and 70°C to displacing oil out of the sample until equilibrium is reached, the volume of the displacing oil is measured.





Figure 3. test tubes that used in this study Figure 4. Vacuum Chamber that used in this study.



Figure 5. Experimental flow chart procedures used in this study

4. Results and Discussion

4.1 GWLI Analyses Results

The average pH value of GWLI is 11.11, the conductivity is 173.5 ms/cm, the salinity is 1.7 ppt, and acidity of GWLI is 0.00235. The densities of GWLI is 1.089 g/ml and distillate water is 0.998 g/ml. The average consuming volume of EDTA solution is 0.375, calcium is 15 mg/l, and the magnesium is 49.8 mg/ L. Table 1 showed the concentration of sulphite (SO₄) in GWLI with change in salinity and pH. The porosity is range from 27% to 14% for carbonate rocks, while it is range from was 31% to 23% for sandstone rocks.



Figure 6. Vacuum system.





Figure 7. Baker in oven Table 1. Sulfate concentration result

Sample GWLI	Y(ABS)	Х
Origin GWLI without change	1.004	91.41667
GWLI with salinity 23	1.84	2094.083
GWLI with salinity 10	4.055	3802.333
GWLI with pH 7	7.52	634.4167
GWLI with pH 4	0.277	1541.667

Figure 8. Close pipe orifice to prevent evaporation of oil

4.2 Results of Spontaneous Imbibtion Test

4.2.1 Results of Carbonate core samples that aged in oil for long time period. Figure 9 shows three carbonate core samples (C211, C212, and C213) with pH is 11.17 and salinity is 170 ppt. At 71:55 hours, cumulative oil recovery at room temperature was reached at (2.24%, 19.84%, 5.11%), respectively. The cumulative oil recovery continues increaseing at different temperature until reach maximum recovery with 70°C and 817:45hr to (53.94%, 53.44%, 51.13%), respectively.

Figure 10 shows carbonate cores (C210 and C214) that imbibition by GWLI at pH=7 and C413 at pH=4 and salinity is 170 ppt. The cumulative oil recovery at room temperature was stable at (0.89%, 9.20%), respectively. The cumulative oil recovery increases gradually with increase the time and temperature until at 70°C to (34.05%, 57.09%) for (C210, C214).

Figure 11 shows the cumulative oil recovery in carbonate core samples ((C414, C412) and (C407, C409)) with salinity are 23ppt and 10ppt and pH is 11.17. At the room temperature and at 169:30hr, the cumulative oil recovery are (3.83%, 0.78%) and (7.59%, 21.88%), respectively. At 70°C at with 1201:00hr, the oil recovery increase spontaneously to 44.10%, 35.13% and (60.74%, 39.15%), respectively. The cumulative oil recovery is increasing with decreasing the salinity. However, companion data sets for water-floods and SIT both showed increased recovery with a decrease in salinity ([17], [18], [19]; [20]).

4.2.2 Sandstone core samples that aged in oil for long time period. Figure 12 shows sandstone core samples (S105, S115, S102), that aged in oil to nearly one year. SIT used with pH is 11.17 and salinity is 170 ppt. At room temperature & 71:55hr, the oil recovery are (6.89%, 5.80%, 5.33%), respectively. The cumulative oil recovery is increases gradually with increase oven temperature from room temperature to (30, 40, 50, 60, and 70°C). At 70°C & 817:45hr, the cumulative oil recovery are (63.05%, 55.16%, 54.41%), respectively. Figure 14 shows the result of cores (S107& S104) with pH is 7 and salinity is 170 ppt, and also shows the results of (S316 & S311) with pH is 4 and salinity is 170 ppt. At 40°C & 286:00hr, the oil recovery is (1.89%, 4.00%) and (0.12%, 2.28%), respectively. At

70°C & 1201:00hr, the cumulative oil recovery reach with the maximum rate to (21.76%, 19.20%) and (24.01%, 21.29%), respectively. Figure 14 shows sandstone core that aged in oil for long time period with change salinity to 23 ppt (S307, S303) and 10 ppt (S116, S114) with constant pH is 11.17. At 216:30hr & 30°C, the cumulative oil recovery of (S116, S114) are (2.85%, 8.54%), respectively. At 169:30hr & room the temperature, the cumulative oil recovery of (S307, S303) are (8.97%, 9.49%), respectively. The cumulative oil recovery continues is increased from 30°C to 70°C to (31.37%, 30.50%) for (S116, S114) and (70.01\%, 56.95\%) for (S307, S303), respectively.



Figure 9. Effect of GWLI on cumulative oil recovery for carbonate aged in oil for long time.



Figure 11. Effect of salinity on cumulative oil recovery for carbonate aged in oil for long time.



Figure 10. Effect of pH on cumulative oil recovery for carbonate core aged in oil for long time.



Figure 12. Effect of GWLI on cumulative oil recovery for sandstone core that aged in oil for long time.

4.2.3 Comparison the effects between oil recoveries (carbonate rocks). Figure 15 shows the comparison between oil recoveries with different carbonate cores, temperature, aging time, pH, and salinity. The cumulative oil recovery is approximate and reasonable between 51% and 53.94% in cores (C211, C212, and C213) with pH is 11.7 and 170 ppt. While pH=7 and pH=4 is different, the cumulative oil recovery in core (C210, C214) with pH is 7 is higher than with pH is 4 (C413). The cumulative oil recovery in cores (C407, C409) with low saintly is higher than that with high salinity.

4.2.4 Comparison the effects between oil recoveries (sandstone rocks). Figure 16 shows the cumulative oil recovery in cores (S105, S115, and S102) with high pH and salinity. The cumulative oil in (S107, S104) with pH is 7 is lower than in (S316, S311) with pH is 4. In core (S303, S307) with high salinity (23 ppt) the cumulative oil recovery is higher than with low salinity for core (S114, S116).

4.2.5 Results of Carbonate core samples that aged in oil for Short time period. Figure 17 shows the oil recovery on carbonate core (C009) with short time, high pH, and salinity. At 46:30hr & room temperature, the oil recovery is (12.11%) and at 50°C & 622:00hr it is 15.14%. After that the cumulative oil recovery increases to 18.16%. At 886:30hr & 70°C, the cumulative oil recovery increase to 60.56%.

ICEMINE	IOP Publishing
IOP Conf. Series: Earth and Environmental Science 212 (2018) 012072	doi:10.1088/1755-1315/212/1/012072

Figure 18 shows the effect of pH is 7 and pH is 4 with constant salinity is 170 ppt on oil recovery. At 160:30hr, the cumulative oil recovery reach to (38.90%, 23.05%) in cores (C002, C006) with pH is 7. While are (3.91%, 26.53%) in (C001, C003) with pH is 4 at 30°C. At 40°C & 309.30hr, the cumulative oil recovery continues increasing in (C002) is (48.24%), and at (51.87%) in (C006) at 500:30hr. At 454:00hr, the cumulative oil recovery reach to (5.86%) in (C001), where it reach to (36.09%), while in core (C003) at 405:30 hours. At 1123:00hr & 70°C, the cumulative oil recovery increase gradually and reached (57.57%, 72.04%) in (C002, C006), and to (39.11%, 48.83%) in core (C001, C003), respectively. Figure 19 shows cores (C004 & C005) with 10 ppt and cores (C008 & C007) with 7 ppt at constant pH is 11.17. At 118:30hr & room temperature, the cumulative oil recovery is (48.06\%, 50.97\%) in (C004, C005). At 260:30hr, the cumulative oil recovery is (23.77\%, 20.56\%) in (C008, C007). At 40°C & 333:30hr, the cumulative oil recovery increase is (40.56\%) in (C008). At 454:00hr, the cumulative oil recovery is (41.12%) in core (C007). At 357:00hr, the cumulative oil recovery is (57.67\%) in core (C004), while is it still stable in core (C005). The cumulative oil recovery is (57.67\%) in (C008, C007, C004, C005), respectively.

Figure 20 shows cores (S009 & S010) with pH is 11 and salinity is 170 ppt for short time. At 118:00hr & room temperature, the cumulative oil recovery is (41.19%, 7.87%) in cores (S009 & S010), respectively. At 40°C & 263:00hr, the cumulative oil recovery increase gradually until reach for core (S009) is (72.31%), and at 360:00hr in (S010) is (28.33%). The cumulative oil recovery is increasing with increasing the temperature to 70°C for (S009, S010) to (81.47%, 64.53%) at 981:00hr, respectively.



Figure 13. Effect of pH on cumulative oil recovery for sandstone core that aged in oil for long time.







Figure 14. Effect of salinity on cumulative oil recovery for sandstone core that aged in oil for long time.



Figure 16. Comparison between oil recoveries for sandstone core that aged for long time.

Figure 21 shows core (S001, S002) with pH=7 and (S006, S004) with pH=4 with high salinity. At room temperature, there was no oil produced from core (S001, S002), while the cumulative oil recovery for core (S006, S004) is (7.82%, 17.00%) at 260:30hr. At 40°C & 407:00hr, the cumulative oil recovery is (21.77%, 3.34%) in core (S001, S002). In core (S006, S004), the cumulative oil recovery increases gradually until reach to (13.69%, 36.55%) with 500:30hr. At 70°C &1123:00hr, the

cumulative oil recovery increase exponentially are (18.58%, 54.40%, 49.60%, 15.88%) in (S006, S004, S001, S002), respectively. Figure 22 shows (S008, S007) with 23 ppt and for (S003, S005) with 10 ppt and constant pH is 11.17 with short time. At:30hr & with room tmperature, the cumulative oil recovery for (S008, S007, S003, S005) are (14.12%, 16.51%, 71.46%, 7.46%), respectively. At 1123hr & 70°C, the cumultive oil recovery for (S008, S007, S003, S005) are (72.97%, 72.86%, 73.3%, 43.52%), respectively.



Figure 17. Effect of GWLI on cumulative oil recovery for carbonate that aged in oil for short time.



Figure 19. Effect of salinity on cumulative oil recovery for carbonate core that aged in oil for short time.



Figure 18. Effect of pH on cumulative oil recovery for carbonate that aged in oil for short time.



Figure 20. Effect of GWLI on cumulative oil recovery for sandstone that aged in oil for short time.

4.2.6 Comparison the effects between oil recoveries (carbonate rock). Figure 23 shows carbonate rocks with salinity (23 ppt, 10 ppt) and acidity (pH=7 & pH=4). The cumulative oil recovery with salinity 23 ppt and 10 ppt are approximate and reasonable between 53.15% and 58.87% for (C008, C007, C004, and C005). The cumulative oil recovery in (C006) with pH=7 is very high compared to (C001, and C003), and they are (57.57%, 72.04%, 39.11%, 48.83%) in cores (C002, C006, C001, C003) respectively. The cumulative oil recovery for core with salinity 10 ppt is higher than 23ppt, they are (58.87%, 54.37%) in (C004, C005), respectively, and to (53.15%, 56.55%) in (C008, C007), respectively.

4.2.7 Comparison the effects between oil recoveries (sandstone rocks). Figure 24 shows cores (S008, S007, S003) with salinity 23 ppt and 10 ppt and the oil recoveries are very high. In core (S006, S004) and (S001, S002) with pH=7 and pH=4, respectively. The oil recovery was very small reach maximum to 45.40% in core sample S006. In cores (S004, S001, and S002) are (18.58%, 4.10%, 15.88%). In core (S009, S010) with high salinity and high pH, the cumulative oil recovery is high reach to maximum recovery.

doi:10.1088/1755-1315/212/1/012072

IOP Conf. Series: Earth and Environmental Science 212 (2018) 012072



Figure 21. Effect of pH on cumulative oil recovery for sandstone core that aged in oil for short time.



Figure 23. Comparison between oil recoveries for carbonate core that aged in oil for short time.



Figure 22. Effect of salinity on the oil recovery for sandstone core that aged in oil for short time.



Figure 24. Comparison the effects between oil recoveries for sandstone core that aged in oil for short time.

Conclusion and Recommendation of the Study

SIT was conducted by GWLI with pH is 7 & 4 and salinity is 23 ppt & 10 ppt for long and short aging time. GWLI acidity (pH) has impact effect on oil recovery, when the pH is 4 the oil recovery is higher than pH is 7. The oil recovery without change the salinity and the acidity of GWLI is higher. The oil recovery is increased with decreased the salinity of GWLI. The oil recovery in sandstone with original properties of GWLI is higher than with change any properties. The aging time has impact effect on oil recovery, it decreased with increasing aging time and increasing with decreased aging time. Hopefully, the research findings shown in this study can possibly be useful for references and for operating companies as an important source for understanding and visualizing the effects of pH, salinity and aging core in oil, on oil recovery from sandstone and carbonate reservoirs using GWLI. SIT shuld have amott cell that provides more accurate results. If the amott cell is not available, then SIT must has wider diameter pipe, in order to let the oil freely from the core samples. The amott cell shuld be in one place and do not move it, only in emergency case.

Acknowledgment

The authors would like to thank the department of petroleum engineering, faculty of mining and energy and department of chemistry, faculty of science, Sebha University, and Mechanical Engineering, Faculty of Engineering, Misurata University, Libya.

References

- [1] Anderson, W. G., 1987. "Wettability Literature Survey Part4: Effect of Wettability on Capillary Pressure". Journal of petroleum technology, October, pp 1283-1300.
- [2] Castor, T. P., Somerton, W. H., and Kelly, J, F., 1981. "Recovery Mechanisms of Alkaline Flooding. In: Surface Phenomena in Enhanced Oil Recovery". Ed. Shah, D. O. Plenum Press, New York City, pp. 249-291.
- [3] Anderson, W. G., 1986b. "Wettability Literature Survey Part2: Wettability Measurements". Journal of petroleum technology, trans., AIME281, pp. 1246-1262.

- [4] Craig, F. F., 1971. "*The Reservoir Engineering Aspects of Water-flooding*", Society of Petroleum Engineers Monograph Series, No. 3, Dallas, Texas.
- [5] Cuiec, L., 1975. "*Restoration of the Natural State of Core Samples*," paper SPE 5634 presented at the SPE Annual Technical Conference and Exhibition, Dallas, Sept. 28-Oct. 1.
- [6] Cuiec, L., 1977. "Study of Problems Related to the Restoration of the Natural State of Core Samples," J. Cdn. Pet. Tech., 16, Oct-Dec, pp. 66-80.
- [7] Dangerfield, J.A., and Brown, D.A., 1985. "*The Ekofisk Field: North Sea Oil and Gas Reservoirs*", J. Kleepe et al. (eds), Graharm and Trotman, London, pp.3-22.
- [8] Farouq-Ali, S. M. and Stahl, C. D., 1970. *"Increased oil recovery by improved water-flooding"*. Earth and Mineral Sciences, Volume 39, No. 4, January, pp. 25-28.
- [9] Handy, L.L., 1960. "Determination of Effective Capillary Pressures for Porous Media from Imbibition Data", Trans. AIME, Vol.80.
- [10] Mungan, N., 1972. "Relative Permeability Measurements Using Reservoir Fluids," Soc. Pet. Eng. J., Oct. pp. 348-402.
- [11] Tang G. and Morrow, N.R., 1997. "Salinity, Temperature, Oil Composition and Oil Recovery by Water-flooding", SPE Reservoir Engineering, pp.269-276.
- [12] Wendel, D. J., Anderson, W. G. And Mayers, J. D., 1987. "Restored-state Core Analysis for the *Hutton Reservoir*," SPE Formation Evaluation, December, pp 509-517.
- [13] George, H, and Danhua L, Z, Rice, U., 2004 "Surface Chemistry of Oil Recovery vfrom Fractured, Oil-Wet, Carbonate Formation". SPE 88365. SPE International Symposium on Oilfield Chemistry, Houston, 5–8 February.
- [14] Madi, A, N. , Mohamed, E. Hesham, A. Muammer, A. Abdalsalam, M., 2018 "Effects of Temperature and Aging Time on Oil Recovery in Carbonate and Sandstone Reservoirs using Sea-Water". First International Conference on Science and Technology., 12th – 14th February 2018
- [15] Madi, A, N,. 2016., "A laboratory investigation of the effects of temperature, hardness, surfactants and alkaline on oil recovery from carbonate reservoirs using spontaneous imbibition tests"., J Fundam Renewable Energy App.
- [16] Austad, T., RezaeiDoust, A. and Puntervold, T., 2010. "Chemical mechanism of low salinity water flooding in sandstone reservoirs". Paper SPE 129767 prepared for presentation at the 2010 SPE Improved Oil Recovery Symposium, 24-28 April.
- [17] Tang, G. and Morrow, N.R., 1999a. "Influence of Brine Composition and Fines Migration on Crude Oil/Brine/Rock Interactions and Oil Recovery" Journal of Petroleum Science and Engineering, 24, Dec. 99-111.
- [18] Tang, G., and Morrow, N.R., 1999b. "Oil Recovery by Waterflooding and Imbibition Invading Brine Cation Valency and Salinity", Proceedings of the International Symposium of the Society of Core Analysts, Golden, CO, August.
- [19] Tang, G. and Morrow, N.R., 2002. "Injection of Dilute Brine and Crude Oil/Brine/Rock Interactions", Environmental Mechanics: Water, Mass and Energy Transfer in the Biosphere, Geophysical Monograph 129, American Geophysical Union, 171-179.
- [20] Zhang, Y. and Morrow, N.R., 2006. "Comparison of Secondary and Tertiary Recovery with Change in Injection Brine Composition for Crude Oil/Sandstone Combinations".