

Available online at http://www.journalcra.com

International Journal of Current Research Vol. 14, Issue, 01, pp.20357-20360, January, 2022 DOI: https://doi.org/10.24941/ijcr.42853.01.2022 INTERNATIONAL JOURNAL OF CURRENT RESEARCH

RESEARCH ARTICLE

DETERMINATION OF SELECTED HEAVY METALS CONCENTRATIONS (Cr, Fe, Zn, Pb, Cu, Ni, AND Ti) IN LAWDAR'S TANNERY-YEMEN

Nasser M. N. Masood¹, Adel A. M. Saeed^{2*}, and Ali N. A. Al-Kumi¹

¹Department of Chemistry/Faculty of Education, University of Abyan, Lawdar, Abyan, Yemen ²Department of Chemistry/Faculty of Science, University of Aden, Khormaksar, Aden, Yemen

ARTICLE INFO

Article History: Received 07th October, 2021 Received in revised form 16th November, 2021 Accepted 14th December, 2021 Published online 28th January, 2022

Keywords:

Tannery Effluent, Heavy Metals, ICP-OES.

*Corresponding author: Adel A. M. Saeed

ABSTRACT

This study has been carried out at Lawdar's tannery (Aden-Yemen), and aimed to determine the mean concentration of seven selected heavy metals (Cr, Fe, Zn, Cu, Pb, Ni, and Ti). four samples (three liquid samples and one solid sample), were analyzed by the ICP-OES technique. The results obtained from this study showed overall mean concentrations of the selected heavy metals in the range of (Cr =0.066 -2.366 mg/L), (Fe=2.230 - 6.519 mg/L), (Zn=4.430 - 42.029 mg/L), (Cu= 0.033 - 3.166 mg/L), (Pb=0.133 - 0.866 mg/L), (Ni =0.033 - 0.699 mg/L), and (Ti=0.266 - 49.49 mg/L), respectively. The mean concentrations of the selected heavy metals in the wastewater were above the recommended limit.

Copyright © 2022. Nasser M. N. Masood et al. This is an open access article distributed under the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

Citation: Nasser M. N. Masood, Adel A. M. Saeed, and Ali N. A. Al-Kumi. "Determination of selected heavy metals concentrations (cr, fe, zn, pb, cu, ni, and ti) in lawdar, s tannery-Yemen", 2022. International Journal of Current Research, 14, (01), 20357-20360.

INTRODUCTION

Tannery waste is generated in huge amounts during the process of tanning by leather industries throughout the world. It has been considered as one of the most polluted industrial wastes that contain high levels of metals which are very toxic to animals, plants, and soil. Tannery wastes are of serious consequence since it has a role in the pollution of freshwater bodies, streams, and land. Tanneries, oil refineries, and metal industries are causing de-pollution of surface and groundwater quality (Raj, 1996). Industrial liquid effluents are one of the principal sources of heavy metal responsible for environmental pollution (Solomon et al., 2015). The wastewater from leather tannery contained high amounts of toxic chemicals, heavy metals, and other undesirable substances. As such, among all the industrial wastes, tannery effluents have drained the highest pollutants (Umar et al., 2017). In the environment, Heavy metals are toxic and resist bio-degradation which are discharging pollutants from industrial wastewater.

Disposal of effluents from the industries has resulted in serious contamination of numerous sites. Elements such as cadmium (Cd), chromium (Cr), copper (Cu), Lead (Pb), zinc (Zn), mercury (Hg), etc. have toxic effects on human health and also non-renewable resources. Contamination of surface water by the presence of heavy metals from the natural ecosystem, as pollutants adsorption onto the soil and then to the aqueous environment, etc., has caused serious problems to human health (Vani, 2012). The heavy metal content of the soil is of major significance in relation to their nutrient status and fertility. Metals such as Zn, Cu, and Se are essential elements for the normal growth of plants and living organisms. However, deficiency or excess of these metals could lead to a number of disorders. Non-essential and/or toxic elements such as Ni, Pb or Cr, may be allowed by the environment in low concentrations, become toxic in higher concentrations (Yargholi, 2008). On the contrary with the previous study (Saeed, 2021), the recent work focused on analysis heavy metals levels (i.e. chromium (Cr), iron (Fe), zinc (Zn), lead (Pb), copper (Cu), nickel (Ni), and titanium (Ti)) in Lawdar's

tannery effluents to check the effect of the hazard on health and environment by comparing the results with some limits standards.

MATERIALS AND METHODS

Description of the study area: The study was conducted on Al-Harib's tannery, Lawdar town, Abyan Governorate. The tannery is located in the heart of the town near Lawdar Old Fuel Station, on the main road leading to Lawdar General Hospital. It is located about 180 Km to the east of Aden, south of Yemen. The tannery area is about $11,000 \text{ m}^2$ and consists of three rooms. The first room is used to gather skins from the market. The second room is for chemical processing such as inorganic tanning etc. The third room is for storing tanned leather to export to local and international markets. After tanning, the waste is disposed to sewage.



Fig. 1. Map of study location

Samples Collection: The wastewater samples of the tannery were collected during the period from October 2019 to June 2021. Four samples were collected. The first two samples were collected from the tank that was devoted to the wastewater of the tannery. The third sample was a mixed sample of the wastewater from the tank and the soil near the sewage in the study area. The fourth sample was taken from the soil near the sewage to the sewers in the study area. These samples were saved in dark polythene bottles and treated as mentioned in (Jeffery, 1996). The investigated heavy metals were determined by the Inductively Coupled Plasma-Optical Emission Spectroscopy (ICP-OES) technique.

Samples Digestion: After collecting the fluid samples, the digestion started in which 50 mL of the sample plus 5mL of concentrated HNO₃ was heated ~4hr in a high-heat resistant beaker using an electrical heater in order to make yellow sediment.

The process was repeated once again and then the sediment was left to get cold then, diluted by redistilled water (~0.1 μ S/cm) up to 50mL solution and became ready to test. For preparing the solid sample, we took one gram from dry soil which was obtained from the soil near of sewage basin in the town by added 15mL of HCl and HNO₃ at ratio 1:3 which was heated on an electrical heater for 4 hr on a thermal plate to make sediment. At that point, redistilled water was added up to 50mL and the solution became ready to test.

Sample Investigation: The analysis process was conducted in the Supreme Board of Drugs and Medical Appliances- Aden, by taking 3mL of digested sample separately and diluting to 100mL. Then, diluted samples were filtered so that the ICP-OES was able to test them accurately.

Reagents and Equipment: All the reagents used were of analytical grade without any further purification.

Statistical Analysis: Statistical analysis was performed using SPSS Statistical Package v20. Analysis was repeated at least three times and data were expressed as mean, standard deviation (\pm SD), relative standard deviation (RSD%), standard error (\pm SR), lower bound-upper bound, minimum-maximum (Min-Max) values. and analysis of variance (ANOVA).

RESULTS AND DISCUSSION

Mean concentrations of the selected heavy metals in the tannery effluent are presented in Table (1) and Figure (2).

Table 1. Mean concentration of the selected heavy metal
analyzed in the samples, (n = 4)

Sample		Parameter					
No.	Metal	Mean \pm SD (mg/L)	Selected				
				wavelength λ (nm)			
1	Cr	0.166 ± 0.0001	2.23	267.716			
	Fe	4.660 ± 0.0007	0.53	238.204			
	Zn	12.166 ± 0.0019	0.52	206.200			
	Cu	0.0990 ± 0.0001	3.78	327.393			
	Pb	0.2000 ± 0.0019	32.62	220.353			
	Ni	0.0999 ± 0.0004	16.83	231.604			
	Ti	0.9665 ± 0.0024	8.26	190.801			
2	Cr	0.066 ± 0.00002	8.59	267.716			
	Fe	2.230 ± 0.0002	0.26	238.204			
	Zn	4.430 ± 0.0007	0.51	206.200			
	Cu	0.033 ± 0.0003	16.81	327.393			
	Pb	0.133 ± 0.0010	27.14	220.353			
	Ni	0.033 ± 22.31	22.31	231.604			
	Ti	0.266 ± 0.0021	26.08	190.801			
3	Cr	0.166 ± 0.0002	2.98	267.716			
	Fe	5.799 ± 0.0011	0.61	238.204			
	Zn	8.760 ± 0.0034	1.28	206.200			
	Cu	0.066 ± 0.0001	3.51	327.393			
	Pb	0.499 ± 0.0004	2.50	220.802			
	Ni	0.066 ± 0.0003	19.43	231.604			
	Ti	0.766 ± 0.0024	10.07	190.801			
4	Cr	2.366 ± 0.0005	0.68	267.716			
	Fe	6.519 ± 0.329	1.68	238.204			
	Zn	42.029 ± 0.0114	0.90	206.200			
	Cu	3.166 ± 0.0012	1.28	327.393			
	Pb	0.866 ± 0.0011	4.17	220.353			
	Ni	0.699 ± 0.0002	0.94	231.604			
	Ti	49.49 ± 0.0084	0.56	190.801			

The mean concentrations of Cr, Fe, Zn, Cu, Pb, Ni, and Ti in the first sample were 0.166, 4.660, 12.166, 0.099, 0.200. 0.100 and 0.967 mg /L, respectively. The mean concentrations of the same elements in sample 2, were 0.066, 2.230, 4.430, 0.033, 0.133, 0.033, and 0.266 mg/L, respectively. For the third sample, the heavy metals had the mean concentrations Cr (0.166mg/L), Fe (5.799mg/L), Zn (8.760mg/L), Cu (0.066mg/L), Pb (0.499mg/L), Ni (0.066mg/L), and Ti (0.766 mg/L). Finally, the mean concentrations of Cr, Fe, Zn, Cu, Pb, Ni, and Ti in the fourth sample were found to be 2.366, 6.519, 42.029, 3.166, 0.866, 0.699, and 49.490 mg/L, respectively. Heavy metals levels in the analyzed samples (1, 2, and 3) were followed the order Zn > Fe> Ti > Pb > Cr > Cu = Ni, and sample 2 showed the lowest values.

Metals*	Mean ± SD*	± SR	95%Confidence Interval for Mean		Minimum	Maximum
			Lower Bound	Upper Bound		
Cr	0.6910 ± 1.1177	0.5588	- 1.0874	2.4694	0.066	2.366
Fe	4.8270 ± 1.8931	0.9466	1.8147	7.8393	2.230	6.519
Zn	16.8495 ± 17.0847	8.5423	- 10.3360	44.0350	4.430	42.029
Cu	0.8467 ± 1.5466	0.7733	- 1.6142	3.3077	0.033	3.166
Pb	0.4248 ± 0.3347	0.1673	- 0.1078	0.9573	0.133	0.866
Ni	0.2275 ± 0.324	0.1618	- 0.2872	0.7422	0.033	0.699
Ti	12.8770 ± 24.424	12.212	- 25.9867	51.7407	0.266	49.49

 Table 2. The overall mean concentrations of the selected heavy metals in all samples

* Recommended limits (mg/L) by WHO (2006) (8): Cr (0.10), Fe (0.30), Zn (0.03), Cu (0.01), Pb (0.05), Ni (0.02), Ti (---).

By USEPA (2010) (9): Cr (0.10), Fe (---), Zn (2.00), Cu (1.00), Pb (0.015), Ni (0.2), Ti (---).



Fig. 2. Mean concentration of the selected heavy metals in samples (1-4)



Fig. 3. The overall mean concentrations of the selected heavy metals in all samples

On the other hand, the highest mean concentrations of heavy metals were coined to sample 4 which varied from the abovementioned samples and followed the pattern Ti > Zn > Fe > Cu > Cr > Pb > Ni. In all samples, the mean concentrations of Cr ranged from 0.066 mg/L to 2.366 mg/L. For Fe, the mean concentrations fall between 2.230 and 6.519 mg/L. However, the mean concentrations of Zn ranged from 4.430 to 42.029 mg/L. Cu mean concentration ranged from 0.033 to 3.166 mg/L while Pb mean concentrations ranged from 0.133 to 0.866 mg/L. Ni mean concentrations (0.033-0.699 mg/L) while Ti mean concentration ranged from 0.266 to 49.49 mg/L. The overall mean concentrations of the selected heavy metals in all samples were found to be in the sequence Zn > Ti > Fe > Cu>Cr > Pb > Ni. (see Table 2 and Figure 3). Bernard and Ogunleye (Bernard, 2015) found that the ranges of mean concentrations for Cr, Pb, Fe, and Cu were 3.33 - 5.79 mg/L, 0.67 - 3.10 mg/L, 3.53 - 8.12 mg/L. and 0.82 - 1.51 mg/L respectively which were higher than the maximum permissible limits of WHO and USEPA. Bahiru in his work (Bahiru et al., 2019) concluded that the mean concentrations of Cr in different samples were 0.2±0.00 - 1.04±0.00 mg/L. For Pb, the concentrations ranged between 1.98±0.04 and 3.11±0.04, mg/L, Fe was between 2.89±0.04 and 5.13±0.06 mg/L, and for Cu found to be between 0.3 ± 0.00 and 0.99 ± 0.06 mg/L. These values are above the permissible limit set by WHO (2006). One study (12) summarized the order of metal mean concentration as Cr (2.007-3.73 mg/L) > Mn (1.421-3.049 mg/L > Zn (0.105-0.934 mg/L) > Cu (0.024-1.906 mg/L) > Cd (0.015 to 0.050 mg /L). The recent work showed that the overall mean concentrations of the selected heavy metals were above the permissible limits, given by WHO and almost USEPA, in all samples. This agreed with Akan with his coworkers (Akan et al., 2019) who found that the mean concentration of metals ions in tannery effluents was very high compared to standards limits. The high concentrations of Zn, Ti, and Fe (Figure 3) in the considered samples may be related to simplicity and cost-effectiveness in using their inorganic salts as tanning agents, sustainable wet-white tanning processes, and the pickling natural leather (Nishad et al., 2013; Kleban, 2006; Ferrer, 2012; Crudu, 2014).

CONCLUSION

The results of this study showed that the mean concentration of the selected heavy metals of the soil sample was higher than that found concerning the fluid samples. The mean concentrations of the select heavy metals in all samples were above the limits set by WHO and USEPA.

REFERENCES

- Akan JC, Abdulrahman FI, Ayodele JT, Ogugbuaja VO 2009. Impact of tannery and textile effluent on the chemical characteristics of Challawa River, Kano State, Nigeria. *Aust J Basic Appl Sci* 3(3):1933–1947.
- Bahiru B.D., Teju E., Kebede T., and Demissie N., 2019. Levels of some toxic heavy metals (Cr, Cd, and Pb) in selected vegetables and soil around the eastern industry zone, central. *African Journal of Agricultural Research Journal of Natural Sciences* 14(2):92-101.
- Bernard E. and Ogunleye A. (2015); Evaluation of tannery effluent content in Kano metropolis, Kano State Nigeria. *International Journal of Physical Sciences*, 10(9): 306-310. DOI: 10.5897/IJPS2014.4240

- Crudu, M., Deselnicu V, Deselnicu D.C., Albu L., 2014. Valorization of titanium metal wastes as tanning agent used in leather industry. Waste Management;1-9. DOI: 10.1016/j.wasman.2013.12.015
- Ferrer J., Riqueme M.E., Segarra V., Galiana M. V., Navarro S. Titanium-tanned leather, 4th International Conference on Advanced Materials and Systems, ICAMS 2012; 1-6.
- Hailu T., Abebaw A. and Minilu D. 2018. Quantification of some Physico-Chemical of wastewater Effluent from Batu Tannery Industry at Different Batch, Addis Ababa, Ethiopia. J. Environ. Anal. Toxico. 8(6) DOI: 10.417222161-0525.1000592
- Jeffery G.H. 1996. Vogel's Textbook of Quantitative Chemical Analysis (5th Ed.) Addison Wesley, Longman Ltd., Harlow, England.
- Kleban, M., (2006); Chrome-free Waterproof Leather. US Patent Appl. 200601151738.
- Nishad N.F., Balaraman M., Raghava R.J., and Balachandran U.N. 2003. Mixed metal tanning using chrome-zinc-silica: A new chrome-saver approach. *JALCA*, Vol. 98: 139-146.
- Raj, E.M Sankaran D.P., Sreenath S.K., Kumaran S., and Mohan N. 1996. Studies on treated effluent characteristics of a few tanneries at Crompet, Madras. Indian. J. Environ. Protect, 16: 252-254.
- Saeed A.A.M., Masood N.M.N., and Al-Kumi A.N.A., 2021. Estimation of some physico-chemical parameters of tannery effluents to surrounding environment (A Yemeni case study), Academic Journal of Research and Scientific Publishing (AJRSP), Vol. 3(30):96-110. DOI.org/10.52132/Ajrsp.e.2021.306

- Solomon, S., Yadessa, C., Girma, T. and Daniel, F. 2015. Heavy Metal Concentrations and Physicochemical Characteristics of Effluent along the Discharge Route from Hawassa Textile Factory, Ethiopia. Journal of Environmental Analytical Toxicology, 5(4): 17.
- Umar, I.M, Ibrahim, M.A, Mustapha, M.B., Mohammed I.B., Tashi Obafemi U.T and Ahmad, G.I. 2017. Physicochemical Analysis and Microbiological Assessment of Tannery Effluent Discharged from Tanneries around Nigeria's Kano Industrial Estates. *Journal of Advances in Microbiology*, 2(1): 1-12.
- United States Environmental Protection Agency (USEPA) 2010. Risk-based concentration table. United States Environmental Protection Agency, Washington, DC, USA.
- Vani Y. S., Sridevil V., and Chandana M.V.V. 2012. A Review on Adsorption of Heavy Metals from Aqueous Solution, *Journal of Chemical, Biological and Physical Sciences*. 2 (3): 1585-1593.
- World Health Organization (WHO) (2006); WHO Guidelines for the Safe use of wastewater, excreta, and greywater. Vol. I: Policy and Regulatory Aspects. Vol. II: Wastewater Use in Agriculture. Vol. III: wastewater and excreta use in aquaculture. Vol. IV: excreta and Greywater use in agriculture. Geneva: World Health.
- Yargholi B., and Azimi A., 2008. Investigation of Cadmium absorption and accumulation in different parts of some vegetables American Eurasian J. Agric. Environ. Sci. 3(3): 357-364.
